The Interplay between Experiment and Simulation for the Design of New Metal-Organic Frameworks

Randall Q. Snurr

Department of Chemical & Biological Engineering Northwestern University, Evanston, IL 60208

http://zeolites.cqe.northwestern.edu



## Metal-Organic Frameworks

- "MOFs"
- Permanently porous, crystalline materials
- Metal or metal oxide nodes connected by organic "linker" molecules
- Large surface areas (up to 7000 m<sup>2</sup>/g) and pore volumes
- Nodes and linkers can be tuned for desired purposes



Mulfort, Farha, Stern, Sarjeant, and Hupp, *J. Am. Chem. Soc.*, 2009. Fahra, Yazaydin, Eryazici, Malliakas, Hauser, Kanatzidis, Nguyen, Snurr, and Hupp, *Nature Chem.*, 2010.

## Molecular Tinker Toys





### Materials Design



http://www.cottinginc.com/pressure-swing-adsorption.html

#### Can tune material properties via synthesis

- pore size
- linker functionality
- open-metal sites
- extraframework cations or anions

#### Can also modify MOFs after their synthesis

## Outline

Molecular-level modeling is playing an important role in the development of new MOFs and in increasing our understanding of their properties.

 $\rightarrow$  Can we design new MOFs on the computer?

- Computer-aided design of new MOFs for
  - natural gas storage
  - hydrogen storage
- High-throughput screening of hypothetical MOFs for natural gas storage

## Molecular Simulation of Adsorption



# Simulation Model

Atomistic representation of MOFs



- MOF atoms are held fixed at their crystallographic coordinates.
- Lennard-Jones parameters taken from the DREIDING force field.
- Charges on framework atoms from quantum chemical calculations.
- Atomistic representation of guest molecules
  - E.g. CO<sub>2</sub>/CO<sub>2</sub> parameters taken from TraPPE force field that matches bulk vapor/liquid equilibria\*
  - Lennard-Jones + Coulomb



\* Potoff, Siepmann, AIChE J., 2001.

## **Molecular Simulation Code**

#### Music – multipurpose simulation code

- Object-oriented F90 code
- Capable of GCMC, MD
- Publicly available from our web site



Gupta, Chempath, Sanborn, Clark, Snurr, *Molecular Simulation*, 2003. Chempath, Düren, Sarkisov, Snurr, *Molecular Simulation*, 2013.

## Predicted CH<sub>4</sub> Adsorption Isotherms



Düren, Sarkisov, Yaghi, Snurr, Langmuir, 2004.

## CO<sub>2</sub> Adsorption in MOFs



NU-111 - CO<sub>2</sub> Absolute Isotherms Simulation vs. Experiment

Walton, Millward, Dubbeldam, Frost, Low, Yaghi, Snurr, *J. Am. Chem. Soc.*, 2008.

Peng, Srinivasa, Wilmer, Eryazici, Snurr, Hupp, Yildirim, Farha, *Chem. Commun.*, 2013.

## Working Hypothesis

Molecular modeling can predict adsorption of small molecules in rigid MOFs that do not have strongly-interacting functional groups in good agreement with experiment.

Challenges:

- Flexible MOFs
- MOFs with interesting functional groups

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## Storage of Natural Gas

**Compressed Natural Gas** 

- rather heavy pressure vessels
- storage at pressure > 207 bar requires expensive multistage compression
- energy density is lower than in gasoline

Alternative: Adsorbed Natural Gas

- storage by physical adsorption at lower pressure
- less expensive compression
- Iower cost of storage vessels

DOE target: 180 cm<sup>3</sup>(STP) / cm<sup>3</sup> at 35 bar ARPA-E target: ~300 cm<sup>3</sup>(STP) / cm<sup>3</sup>

## Predicted CH<sub>4</sub> Adsorption Isotherms



Why is the methane uptake so high?

Düren, Sarkisov, Yaghi, Snurr, Langmuir, 2004.

## Comparison with Other Adsorbents

## Molecular squares

pyrazine squares bipyridine squares porphyrin squares





## Zeolites

silicalite faujasite









## Carbon nanotubes (CNT)

with and without interstitial adsorption



# Methane Storage

- Optimal material for CH<sub>4</sub> storage should have
  - large accessible surface area
  - high free volume (large capacity)
  - high heat of adsorption
  - narrow pore size distribution
  - low crystalline density (if high storage per gram is important)
- GCMC simulations can reveal the complex interplay of properties responsible for CH<sub>4</sub> adsorption.
- Using these insights, GCMC simulations can be used to screen new candidate materials before their synthesis...

# Rational Design of New IRMOFs



at 35 bar	IRMOF-6	IRMOF-TBrB	IRMOF-Anthracene
N / cm <sup>3</sup> (STP) cm <sup>-3</sup>	135.5	167.2	181.0

# PCN-14



- PCN-13
- Limited methane uptake
- Record methane uptake

**PCN-14** 

Ma, Sun, Simmons, Collier, Yuan, Zhou, J. Am. Chem. Soc., 2008

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## From Modeling to Reality



Farha, Yazaydin, Eryazici, Malliakas, Hauser, Kanatzidis, Nguyen, Snurr, Hupp, Nature Chem., 2010.

#### High Surface Area MOFs: Team Approach Ozqur **Snurr Group Structure prediction** Adsorption simulations Omar Christos Ibrahim Hupp Group **Kanatzidis Group Nguyen Group** Ligand design **Experimental** Ligand synthesis Ligand synthesis structure **MOF** synthesis determination **Physical measurements Brad Hupp Group Collaborators: Physical** Prof. Joe Hupp ulletmeasurements Prof. SonBinh Nguyen ٠ Prof. Mercouri Kanatzidis • **Prof. Omar Farha** ٠

## Computational Design of a High Surface Area MOF



Pressure (bar)

Kanatzidis, Nguyen, Snurr, Hupp, Nature Chem., 2010.

### CO<sub>2</sub> Adsorption



### Hydrogen Storage



28 g/L

DOE targets at 100 bar and ambient temperature: 5.5 wt% 40 g/L

45 g/L

These excess values are the highest reported to date at 77 K.

#### Room Temperature Storage?

Enthalpy of adsorption for  $H_2$  in NU-100 is ~ 6 kJ/mol: weak physisorption.

For room temperature storage, we need ~20 kJ/mol.



### **Improved Adsorption Enthalpies**

Computational exploration of metal-alkoxides

- Li-alkoxide MOFs do not meet DOE hydrogen storage targets.
- Divalent cations have larger electric fields than Li.
- Transition metals can interact strongly with hydrogen.



Getman, Miller, Wang, Snurr, *J. Phys. Chem. C*, 2011. Brand, Colon, Getman, Snurr, *Micropor. Mesopor. Mater.*, 2013.

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### Molecular Tinker Toys



How can we more rapidly find the best MOFs for a given application?

# Virtual High-Throughput Screening

Crystal generator for hypothetical MOFs

- Systematically generates MOF structures from a library of building blocks
- Created a large database of hypothetical MOFs (over 137,000 entries and growing)
- Designed for high-throughput screening of physical properties



Wilmer, Leaf, Lee, Farha, Hauser, Hupp, Snurr, *Nature Chem.*, 2012.

## Virtual High-Throughput Screening





#### Continued...









+ functional groups...



## Database of Hypothetical MOFs

Restricting ourselves to MOFs with

- one type of node and
- one or two types of linkers

 $\rightarrow$  137,953 MOFs

### Finding Improved Methane Storage Materials



Wilmer, Leaf, Lee, Farha, Hauser, Hupp, Snurr, Nature Chem., 2012.



### Structure-Property Relationships



#### **Quantitative Structure-Property Relationships**



Large-scale Quantitative Structure-Property Relationship (QSPR) Analysis of Methane Storage in Metal-organic Frameworks.M. Fernandez, T.K. Woo, C.E. Wilmer, R.Q. Snurr, J. Phys. Chem. C 2013.

#### hmofs.northwestern.edu



#### Search Criteria:

Excess Methane Adsorption<sup>a</sup> , Surface Area<sup>b</sup>, Density<sup>c</sup>, Void Fraction<sup>d</sup>, Pore Size<sup>e</sup>

a Volumetric (cm3(STP)/cm3) greater than	but less than	(Min: 5, Max: 238, either field may be left blan	k)
Gravimetric $(cm^3(STP)/g)$ greater than	but less than	(Min: 1, Max: 435, either field may be left blan	k)



Accessed by researchers from around the world.

### Framework Topologies in the Database



Sikora, Winnegar, Proserpio, Snurr, Microporous Mesoporous Mater., 2014.



#### "Top-down" MOF construction

Selection of appropriate net template

Scaling of unit cell based on targeted building blocks

Placement of inorganic "nodes"

Placement of organic "nodes"

Placement of organic "linkers"

Classical force field optimization of hypothetical structures

Characterization of textural properties

Simulation of methane adsorption isotherm up to 65 bar



**D.A. Gomez-Gualdron**, O. V. Gutov, V. Krungleviciute, B. Borah, J.E. Mondloch, J.T. Hupp, T. Yildirim, O.K. Farha, R.Q. Snurr, Computational design of metal-organic frameworks based on stable zirconium building units for storage and delivery of methane, *Chem. Mater.*, in press.



#### Validation of simulation: methane adsorption

Ratio between simulated deliverable capacity (~190 cc(STP)/cc) and measured one (~170 cc(STP)/cc) is 0.895, which correlates well with the 0.884 ratio between simulated and measured BET surface area



**D.A. Gomez-Gualdron**, O. V. Gutov, V. Krungleviciute, B. Borah, J.E. Mondloch, J.T. Hupp, T. Yildirim, O.K. Farha, R.Q. Snurr, Computational design of metal-organic frameworks based on stable zirconium building units for storage and delivery of methane, *Chem. Mater.*, in press.

#### Computation-Ready, Experimental (CoRE) MOF Database



- MOF structures identified in the CSD
- Automated routines developed to "clean up" original crystal structures
  - Solvents (free & coordinated) are removed
  - lons are retained
  - Structures with missing hydrogen atoms are identified and fixed
  - Mislabeled structures are recovered by text-mining
- >5,100 "computation-ready" structures with PLD > 2.4 Å

Chung, Camp, Haranczyk, Sikora, Bury, Krungleviciute, Yildirim, Farha, Sholl, Snurr, Chem. Mater., in press.



## Conclusions

How close are we to designing MOFs on the computer?

- Molecular simulation can provide (and has provided) inspiration for new materials.
- Proposed MOFs can be evaluated readily on the computer for their ability to store and separate simple gases.
- High-throughput computational screening is still in its infancy. It requires
  - Generation tools
  - Methods to calculate properties
  - Metrics for evaluation
  - Data mining

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